A FORMAL, STEREOCONTROLLED SYNTHESIS OF (±) ESTRONE EMPLOYING THE TRIMETHYLSILYLCYANOHYDRIN COPE REARRANGEMENT

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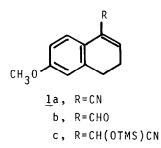
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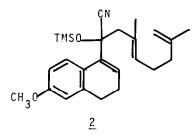
Summary: A new synthetic approach to estrone is discussed which employs control of stereochemistry via the Cope rearrangement.

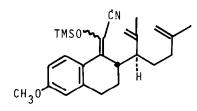
Steroids, by virtue of their polycyclic structures and well defined stereochemistries, have provided a means for the development of new synthetic strategies and the study of the stereochemical course of chemical reactions. Among the steroids, estrogens have served this role most admirably.¹ We have recently reported² a Cope rearrangement which provides $\delta_{,\varepsilon}$ unsaturated esters. In order to explore the stereochemical course of this rearrangement, we have chosen estrone as a synthetic target and, in so doing, have also developed a polyene approach to the steroid nucleus which introduces an 11β -hydroxyl function rendering the two methods applicable to the synthesis of steroids.

The cyanohydrin of 6-methoxy-1-tetralone, prepared via the TMSCN method,³ was dehydrated as described by Nagata⁴ to provide unsaturated nitrile <u>1a</u> (77%). Reduction of the nitrile (DIBAL, $-78^{\circ}C \rightarrow 0^{\circ}C$, ether-hexane) afforded the unsaturated aldehyde⁵ (85%) which was readily converted (TMSCN, ZnI₂ cat., CH₂Cl₂, 41°C, 24 h) to its trimethylsilylcyanohydrin <u>1c</u>. The lithium anion of nitrile <u>1c</u> (n-BuLi, THF, $-78^{\circ}C$) was alkylated ($-78^{\circ}C \rightarrow 25^{\circ}C$, 18 h) with 1-chloro-2,6-dimethyl-2(E),6-heptadiene⁶ giving rise to the aryltriene <u>2</u> as a yellow liquid (58%).

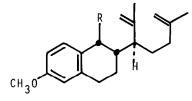
Upon heating $(160^{\circ}C, 50 h, N_2)$ a solution of <u>2</u> in dimethylaniline, Cope rearrangement to the cinnamonitriles <u>3</u> was effected. The Cope rearrangement of the pro-homo-D-ring was not anticipated at such a low temperature.⁷ Exposure of <u>3</u> to KF in refluxing methanol afforded (67% from <u>2</u>) a pair of diastereometric esters <u>4a</u> and <u>5a</u> in an 80/20 ratio, respectively. The major isomer was tentatively assigned the cis stereochemistry since equilibration (NaOCH₃/HOCH₃, reflux, 19 h) provided <u>4a/5a</u> in a 17/83 ratio. Each isomer displayed 22 signals in its ¹³C spectrum, attesting to its stereoisometric purity. The stereochemistry at



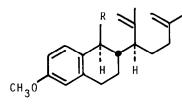


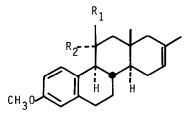


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<u>4</u>a, R=CO₂CH₃ b, R=CH₂OH c, R=CHO





5a, R=C0₂CH₃ b, R=CH₂OH c, R=CH0

 $\begin{array}{c} \underline{6}a, \ R_{1}=0H, \ R_{2}=H\\ b, \ R_{1}=H, \ R_{2}=0H\\ c, \ R_{1}=0Ms, \ R_{2}=H\\ d, \ R_{1}=H, \ R_{2}=0Ms\\ e, \ R_{1}, R_{2}=0\\ f, \ R_{1}=R_{2}=H \end{array}$

 $pro-C_8$ and $pro-C_{14}$ (steroid numbering) was tentatively assigned as depicted, arising from a chair-like transition state.

The individual esters were reduced (LiAlH₄, ether) to their respective alcohols (<u>4b/5b</u>) and subsequently oxidized (PCC, NaOAc, CH_2Cl_2)⁸ to their aldehydes <u>4c</u> and <u>5c</u>. Exposure of either aldehyde to NaOCH₃/HOCH₃ resulted in a 92/8 (<u>5c/4c</u>) equilibrium mixture. Both aldehydes, upon irradiation of the aldehyde proton, revealed the pro-C₉ proton as a doublet (J = 3 Hz) which precluded definitive assignment of their stereochemistries on the basis of coupling constants. The stereochemical relationship between pro-C₈ and pro-C₉ was answered in the ensuing transformation.

Aldehyde <u>5c</u> was readily cyclized $(SnCl_4/CH_3NO_2, 0^{\circ}C, 3 \min)^9$ to the tetracyclic alcohol <u>6a</u> in 50% yield. No other identifiable products could be isolated. The relative stereochemistry at C₈, C₉, and C₁₁ of <u>6a</u> was readily apparent upon examination of its 270 MHz nmr spectrum. The C₉ proton appeared at δ 2.55 (dd, J_{8,9} = 11 Hz, J_{9,11} = 3 Hz) requiring a trans-B,C juncture while the C₁₁ proton at δ 4.69 (q, J = 3 Hz) necessitated an axial hydroxyl group. Confirmation of the stereochemistry of the remaining centers was achieved by removal of the C₁₁ hydroxyl function.

Attempted mesylation (MsCl, Et₃N, CH₂Cl₂, 0°C)⁹ of <u>6a</u> produced a mixture of olefins $(\Delta^{9,10}, \Delta^{10,11})$ presumably arising from facile trans diaxial elimination of the expected mesylate. However, oxidation (PDC, CH₂Cl₂, 25°C, 10 h) of axial alcohol <u>6a</u> provided ketone <u>6e</u>, mp 161-163°C (60%) (C₉-H, δ 3.68, d, J = 11 Hz) which was subsequently reduced (Li/NH₃/THF) to provide the equatorial alcohol <u>6b</u> [C₁₁-H, 4.14 (td, J = 11 and 3 Hz)]. Mesylation of <u>6b</u> proceeded without incident, producing <u>6d</u> which was reduced (Li/NH₃/THF) to the crystal-line tetracyclic <u>6f</u>, mp 96.5-97.5°C (lit.^{1g} 96-97°C) whose 270 MHz nmr spectrum was identical with an authentic sample.

The synthesis of <u>6f</u> confirms the chair-like transition state of the Cope rearrangement and provides a formal total synthesis of estrone, since <u>6f</u> has previously been converted to estrone by Valenta and co-workers.^{1g} Moreover, <u>6f</u> provides a potential intermediate for syntheses of corticoid steroids.¹¹

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REFERENCES AND NOTES:

- For a review, see D. Taub, "Naturally Occurring Aromatic Steroids," in <u>The Total</u> <u>Synthesis</u> of Natural Products, John ApSimon, ed. Vol 2, p 641. John Wiley, (1973). For more recent contributions, see, a) P. A. Bartlett and W. S. Johnson, J. Am. Chem. Soc., <u>95</u>, 7501 (1973); b) N. Cohen, B. L. Banner, W. F. Eichel, D. R. Parrish, G. Saucy, J.-M. Cassal, W. Meier, and A. Fuerst, J. Org. Chem., <u>40</u>, 681 (1975); c) U. Eder, H. Gibian, G. Haffer, G. Neef, G. Sauer, and R. Wiechert, Chem. Ber., <u>109</u>, 2948 (1976); d) S. Danishefsky and P. Cain, J. Am. Chem. Soc., <u>98</u>, 4975 (1976); e) T. Kametani, H. Nemoto, H. Ishikawa, K. Shiroyama, H. Matsumoto, and K. Fukunoto, <u>ibid.</u>, <u>99</u>, 3461 (1977); f) W. Oppolzer, K. Baettig, and M. Petrzilka, Helv. Chim. Acta, <u>61</u>, 1945 (1978); g) J. Das, R. Kubela, G. A. MacAlpine, Z. Stojanac and Z. Valenta, Can. J. Chem., <u>57</u>, 3308 (1979); h) T. A. Bryson and C. J. Reichel, Tetrahedron Lett., 2381 (1980); i) P. A. Grieco, T. Takagawa, and W. J. Schillinger, J. Org. Chem., <u>45</u>, 2247 (1980); j) R. L. Funk and K. P. C. Vollhardt, J. Am. Chem. Soc., <u>102</u>, 5253 (1980); k) S. Djuric, T. Sarkar, and P. Magnus, <u>ibid</u>., <u>102</u>, 6886 (1980).
- 2. F. E. Ziegler, R. V. Nelson and T.-F. Wang, Tetrahedron Lett., 2125 (1980).
- D. A. Evans and L. K. Truesdale, Tetrahedron Lett., 4929 (1973); P. G. Gassman and J. J. Talley, ibid., 3773 (1978).
- 4. W. Nagata, M. Yoshioka, and M. Murakami, Org. Synth., Vol 52, p 96, John Wiley, (1972).
- 5. All new compounds gave correct spectroscopic and/or combustion analysis data.
- 6. The chloride was prepared (MsCl, Et₃N, CH₂Cl₂, 3 h) from the alcohol, O. P. Vig, B. Vig, and R. C. Anand, Indian J. Chem., <u>7</u>, 1111 (1969). The Claisen rearrangement was performed at 165°C for 18 h.
- S. J. Rhoads and N. R. Raulins, "The Claisen and Cope Rearrangements," in Org. React., Vol 22, John Wiley (1974).
- 8. E. J. Corey and J. W. Suggs, Tetrahedron Lett., 2647 (1975).
- For biannular cyclization of a diene aldehyde, see R. E. Ireland, M. I. Dawson, C. J. Kowalski, C. A. Lipinski, D. R. Marshall, J. W. Tilley, J. Bordner, and B. L. Trus, J. Org. Chem., <u>40</u>, 973 (1975); For a general review of polyene cyclization, see, W. S. Johnson, Bioorg. Chem., 5, 51 (1976).
- 10. R. K. Crossland and K. C. Servis, J. Org. Chem., <u>35</u>, 3195 (1970).
- 11. P. Turnbull, K. Syhora, and J. H. Fried, J. Am. Chem. Soc., <u>88</u>, 4764 (1966).

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